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Reg. No. :

Name :

Fourth Semester B.Sc. Degree Examination, March 2020

First Degree Programme under CBCSS

Complementary Course

PY 1431.2 – ATOMIC PHYSICS, QUANTUM MECHANICS AND ELECTRONICS

(For Chemistry)

(2018 Admission)

Time : 3 Hours

Max. Marks: 80

SECTION - A

Answer ALL questions in one or two sentences. Each question carries 1 mark

- 1. State Paul's exclusion principle.
- 2. Define magnetic moment of orbital electrons.
- 3. Define isotope effect.
- Define superconductivity.
- 5. Define probability density.
- 6. Define emission spectra.
- What type of source is used to produce microwave radiations.
- 8. What are the materials of the prism used in ultraviolet spectrometers.
- Define ripple factor.
- 10. Write the truth table of AND gate.

(10 × 1 = 10 Marks)

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SECTION-B

Answer any eight questions, not exceeding a paragraph. Each question carries 2 marks

- 11. Briefly explain the quantum numbers associated with vector atom model.
- 12. Explain the various Coupling schemes.
- Discuss briefly the modern periodic table
- 14. Distinguish between type I and type II superconductors.
- 15. Briefly discuss the limitations of classical physics.
- Explain the physical significance of wave function.
- 17. State and explain Planck's hypothesis.
- Write the Schrodinger time-independent equation.
- 19. Briefly discuss the principle of spectrometer used in the infrared region.
- 20. Distinguish between zener breakdown and avalanche breakdown.
- 21. Distinguish between ac and dc load lines.
- 22. Define NAND operator and write its troth table.

 $(8 \times 2 = 16 \text{ Marks})$

SECTION-C

Answer any six questions, Each question carries four marks

- Find the longest wavelength limit of Balmer series, given Rydberg constant as 1.097 × 10⁷ m⁻¹.
- Calculate the frequency of revolution of the electron of the Bohr hydrogen atom in its ground state.

 $E_0 = 8.85 \times 10^{-12} C^2 N^{-1} M^{-2}$ $h = 6.625 \times 10^{-34} J$

- 25. An electron typically spends about 10 *s in an excited state before it drops to a lower state by emitting a photon. How many revolutions does an electron in an n=2 Bohr orbit make in 10 *s?
- 26. A particle limited to the x axis has the wavefunction $\Psi = ax$ between x=0 and x=1; $\Psi = 0$ elsewhere. Find the probability that the particle can be found between x=0.45 and x=0.55.
- 27. For an electron in a one dimensional infinite potential well of width 1A°, Calculate the separation between the two lowest energy levels.
- 28. An ac supply of 230 V is applied to a half-wave rectifier circuit through a transformer of turn ratio 10:1.Find (i) the output dc voltage and (ii) peak inverse voltage. Assume the diode to be ideal.
- 29. The zener diode has $V_z = 18V$. The voltage across the load stays at 18V as long as I_z maintained between 200mA and 2A. Find the value of series resistance R so that E_0 remains 18V while input voltage E_1 is free to vary between 22V and 28V.
- 30. If the collector current changes from 2mA to 3mA in a transistor when collectoremitter voltage is increased from 2V to 10V, what is the output resistance?
- 31. Find the 2's complement of
 - (a) 1001₂
 - (b) 1110₂
 - (c) 100₂

(6 × 4 = 24 Marks)

SECTION-D

Answer any two questions, Each question carries 15 marks

- 32. (a) State the important postulates of Bohr atom model. What are its limitations?
 - (b) Derive an expression for the energy of the electron in the nth orbit of hydrogen atom.

- Using Schrodinger wave equation for a particle in a potential box, obtain the eigen functions and eigen values.
- Explain the rectifying action of a pn junction diode. With the help of a neat circuit diagram, explain the working of a full wave bridge rectifier.
- 35. With the help of a neat circuit diagram, explain the working of a single stage transistor amplifier. Obtain expression for its current gain, voltage gain and power gain.

(2 × 15 = 30 Marks)